REPORT DOC	Form Approved _MB No. J704-0168			
Numer report in training and income statement and common statement and c				ruktinns, väärdnind dikist nid data kourk ek rosid eksimatib on invisit eksi wotti ni ni rosin kristok eligi eksiptiti – juli tirak rise va kvendoton, uluk 1503
1. VGENCY USE ONLY Leave Dianki	L FORT TATE June 1, 1995	Tech Rpt.	D DATES	COVERED
4. TILE AND SUBTITUE Crystal Structure and of Ce ₈ Pd ₂₄ Sb 5. AUTHORIS) Robert A. Gordon and		in an	N000	DING HUMBERS 14-93-1-0904 John Pazik 3134037ess08
7. PERFORMING ORGANIZATION NAME() Department of Chemis Cornell University Ithaca, NY 14853-13	stry			DRMING ORGANIZATION RT NUMBER
2. SPONSORING/MONITORING AGENCY Office of Naval Rese Chemistry Program 800 N. Quincy St. Alexandria, VA 2221 11. SUPPLEMENTARY NOTES	earch EL	TIC ECTE BBJ 1995		NSORING MONITORING NCY REPORT NUMBER
This document has be release and sale; it unlimited	een approved for	public	12b. OIS	TRIBUTION CODE
The ternary compute intermediate vale a cubic cell with a = on 1453 reflections type is composed of with the perovskite-I Fitting the magnetic expression yielded a and an effective high indicating trivalent observed above 3K.	8.461(1)Å, Pm3m (222 unique) and distorted perovslike units center susceptibility of Weiss constant of temperature mon	symmetry with 16 parameter with and Cu ₃ A red on the colata above 10 of -15(3)K (and the per ceri	crysta h wR2 cs. I Au sub crners OK to nti-f	al study yielded = 0.0412 based This new structur ocells arranged s of the cube. a Curie-Weiss erromagnetic) 2.45(4) _{UB}
		DAIG CAVILLA I	ņspeut	ED 3
14. SUBJECT TERMS	crystal structu	re magnetic		15. NUMBER OF PAGES 20

18. SECURITY CLASSIFICATION OF THIS PAGE

Unclassified

Standard Form 298 (Rev. 2-89) Prescribed by ANSI 5td. 239-18 298-102

20. LIMITATION OF ABSTRACT

16. PRICE CODE

19. SECURITY CLASSIFICATION OF ABSTRACT

Unclassified

susceptibility

17. SECURITY CLASSIFICATION OF REPORT

Unclassified

OFFICE OF NAVAL RESEARCH

Grant or Contract N00014-93-1-0904

R&T Code 3134037ess08 Scientific Officer: Dr. John Pazik

Technical Report No. 23

"Crystal Structure and Magnetic Susceptibility of Ce₈Pd₂₄Sb"

b y

Robert A. Gordon and Francis J. DiSalvo

Submitted to

Z. Naturforsch. B

Cornell University
Department of Chemistry
Ithaca, NY 14853

June 1, 1995

Reproduction in whole or in part is permitted for any purpose of the United States Government

This document has been approved for public release and sale; its distribution is unlimited

Crystal Structure and Magnetic Susceptibility of Ce₈Pd₂₄Sb

Robert. A. Gordon and Francis. J. DiSalvo¹
Department of Chemistry,
Cornell University.
Ithaca, NY 14853-1301

Abstract

The ternary compound $Ce_8Pd_{24}Sb$ is very close in composition to the intermediate valent binary $CePd_3$. A single crystal study yielded a cubic cell with a=8.461(1)Å, Pm3m symmetry with wR2 = 0.0412 based on 1453 reflections (222 unique) and 16 parameters. This new structure type is composed of distorted perovskite and Cu_3Au subcells arranged with the perovskite-like units centred on the corners of the cube. Fitting the magnetic susceptibility data above 100K to a Curie-Weiss expression yielded a Weiss constant of -15(3)K (antiferromagnetic) and an effective high temperature moment per cerium of $2.45(4)\mu_B$ indicating trivalent behavior of the cerium atoms. No ordering was observed above 3K.

Keywords: cerium intermetallic, crystal structure, magnetic susceptibility

¹ corresponding author

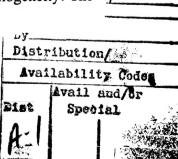
1. Introduction

Interaction between conduction electrons and the 4f electron on cerium can produce unusual electronic and magnetic effects such as magnetic ordering, heavy fermion behavior or intermediate valence (IV), where the 4f electron has significantly but not entirely delocalised [1-2]. A prototypical example of an intermediate valent material is CePd₃, where a strong Kondo interaction gives rise to broad maxima in the Seebeck co-efficient, resistivity and magnetic susceptibility [3-5]. When small elements such as B[3.6] or Si [3,5] were incorporated into CePd₃, in a perovskite-like fashion, the IV state was lost with heavy fermion or magnetic ordering [7] the end result. There is no published report of the incorporation of a larger p-block element into the CePd₃ structure, until now.

We recently began exploring the Pd-rich region of the Ce-Pd-Sb phase diagram to see if new materials could be found which exhibit intermediate valent or heavy fermion behavior. We recently reported the anti-ferromagnet Ce₃Pd₆Sb₅ [8]. Of the 3 other reported ternaries in this system, CePdSb [9], CePd₂Sb₂ [10] and CePdSb₂ [11], only CePdSb is reported to exhibit some Kondo behavior. In discovering the new phase Ce₈Pd₂₄Sb, we are presented with another opportunity to examine the sensitivity of the IV state with respect to small changes in local crystal and electronic structure.

2. Experimental

A sample of composition "Ce₄Pd₁₃Sb_{3.03}" was prepared by arc-melting of the elements, all of at least 99.9% purity. The bead was turned over several times to ensure homogeneity. The



final mass of the bead was consistent with losing 0.03 Sb (1% excess added to compensate for losses due to high Sb vapour pressure). This bead was subsequently placed in a vitreous carbon crucible (EMC Industries, type GAZ-02) and sealed under vacuum in a quartz tube. The tube was then placed within the coil of an induction furnace (Ameritherm Inc.) operating at 195kHz and subjected to the following heat treatment. The sample was first re-melted, then cooled to what appeared, by visual inspection, to be just below the melting point. The sample was then annealed at this point, which, by optical pyrometry, was at a temperature of 1100°C, for 20 hours before being allowed to cool to room temperature. The bead was removed from the tube and cooled to 77K in the hopes that the hard bead would be more brittle at liquid nitrogen temperature. The bead was broken apart under liquid nitrogen in an alumina mortar. After the first break, it became apparent that a single crystalline region approximately 1.5mm thick was present at the bottom of the bead. Further breaks enabled the isolation of both single crystals and multi-crystalline fragments. Inspection of the vitreous carbon crucible under a 40x microscope revealed no evidence of attack of the crucible. The crystals isolated are air-stable and jet black in colour with a mirror-like finish. Following the single crystal study, a sample of the title composition was prepared by arc-melting as above, placed in a section of tantalum tubing and sealed under vacuum in quartz. This bead was annealed for 2 weeks at 900°C.

Precession photographs of a single crystal needle mounted on a glass fibre were taken using an Enraf-Nonius Diffractis 601 and zirconium-filtered molybdenum $K\alpha$ ($\lambda = 0.71073\text{Å}$) radiation. Single crystal intensity data for this same crystal were collected on a Siemans P4 automatic 4-circle diffractometer using graphite monochromatized molybdenum $K\alpha$ radiation and controlled by the XSCAnS program [12]. Powder diffraction data was collected on a SCINTAG

 θ -2 θ Diffractometer using Cu K α_1 radiation. Powder diffraction data on the sample of composition Ce₈Pd₂₄Sb was analysed using the program TREOR [13] and the resulting cell parameters were refined by a least-squares technique. Using the LAZY-PULVERIX program [14], a theoretical pattern was calculated for comparison to the observed powder pattern. Magnetic measurements were performed by the Faraday technique on polycrystalline material from the sample prepared at the title stoichiometry. Magnetic susceptibility data was fit to a Curie-Weiss expression as described previously [15].

3. Results and Discussion

3.1. Crystal Structure

Precession photographs revealed strong peaks that indicated a cubic cell with a lattice constant of 4.2Å but the presence of weak reflections corresponding to twice this lattice parameter indicated the existence of a cubic supercell. No systematic zeros were observed in the photographs, leading to possible space groups P23, Pm3, P432, P43m and Pm3m. Consideration of additional symmetry in the photographs suggested a Lauë symmetry of m3m. The Lauë subroutine of the XSCANS program also indicated m3m symmetry which excludes the first 2 possible space groups. Refinements were carried out in Pm3m (No. 221), which is also the symmetry of CePd₃. Crystallographic data and details of the data collection are summarised in Table 1.

The starting atomic parameters were deduced from an interpretation of direct methods [16] and the structure was then successfully refined using SHELXL-93 [17] with anisotropic displacement parameters for all atoms. Initially, only the Ce and Pd positions were determined, with the Sb position becoming apparent from the difference Fourier map. A final difference Fourier syntheses revealed no significant residual peaks (all between +1.29 and -1.17 e⁻/Å³). The refined atomic parameters and the interatomic distances are listed in Tables 2 and 3. The lattice parameter obtained from a least-squares refinement using all powder diffraction peaks is 8.4445(8)Å. The small difference between powder and single crystal cell size is presumably due to alignment discrepancies between the two diffractometers. A comparison of experimental and calculated powder intensities is given in Table 4.

The structure of Ce₈Pd₂₄Sb (fig. 1) is of a new type, being composed of distorted perovskite and Cu₃Au subcells. The large size of the Sb atom is presumed to be driving the distortion since only an expansion of the lattice has been observed when a small atom such as B occupies the body-centre interstitial site in CePd₃ [3,5,6]. Figure 2a shows the distorted perovskite unit "CeSbPd₃" with an adjoining distorted Cu₃Au unit. The Cu₃Au unit found at the centre of the unit cell (fig. 2b) is essentially an expanded CePd₃ cell 4.207(1)Å on edge (versus 4.126Å in binary CePd₃ [3]), which is comparable to what Kappler et al.[3] refer to as "magnetic CePd₃".

It is evident from fig. 2a that the Sb atom has pushed the Pd(2) atoms outward from the face of the perovskite cell. A close Pd(2)-Sb distance of 2.633(1)Å indicates a strong interaction between these two types of atoms when compared to the sum of their covalent radii, 2.68Å. Such close Pd-Sb contact has also been observed in Ce₃Pd₆Sb₅ (2.629Å) [8] and in EuPd₂Sb₂

(2.601Å) [10]. All other near-neighbour distances are comparable or larger than the sum of the appropriate covalent radii. The Ce-Pd(1) and Ce-Pd(3) distances are both comparable to the Ce-Pd distance in CePd₃ (2.97Å) while the Ce-Pd(2) distance is some 2.5% larger due to the displacement of Pd(2) from the face of the perovskite sub-cell. The Ce-Ce distance is also larger, being 4.207Å between nearest cerium atoms within the cell and 4.254Å between adjacent cells which are 2-3% increases over that in CePd₃ (4.126Å[3]). These Ce-Ce distances are comparable to the limiting cell size when B, Be, Si or Ge [3] are incorporated into the centre of a CePd₃ cell. It is also interesting to note that, while in CePd₃ all Pd-Pd near-neighbour distances are 2.97Å, in Ce₈Pd₂₄Sb, the Pd(2)-Pd(3) and Pd(3)-Pd(3) distances are significantly shortened to 2.765(1)Å and 2.791Å respectively. Both of these distances are comparable to the Pd-Pd distance in palladium metal (2.751Å) [18]. Overall, the structure of Ce₈Pd₂₄Sb could be viewed as a perturbation on CePd₃.

3.2. Magnetic Susceptibility

The inverse of the magnetic susceptibility from 3.0K to 320K is shown in figure 3. No ordering is observed down to 3K but analysis of the data from 100K - 320K indicates an antiferromagnetic Weiss constant of -15(3)K and an effective cerium moment of $2.45(4)\mu_{\rm B}$. Deviations from Curie behavior below 100K may be due to crystal field splitting of the cerium $4f^1$ level, resulting in some temperature dependence to the cerium moment. Additional changes in slope below 50K may indicate the onset of significant magnetic fluctuations. Since the Weiss constant derived from high temperature data is the result of both exchange and crystal field effects [15], it is not possible to simply extract the exchange energy. In contrast to CePd₃, the

effective cerium moment in Ce₈Pd₂₄Sb is consistent with tri-valent behavior. No feature that could be associated with a strong Kondo interaction is apparent. In perturbing CePd₃ by incorporating one antimony per 8 CePd₃ units, with the accompanying small distortions, the intermediate valent state is lost.

4. Conclusions

We have prepared and isolated crystals of a new ternary close in composition to the intermediate valent material CePd₃. This new ternary, Ce₈Pd₂₄Sb, is a cubic compound composed of expanded CePd₃ units and distorted "CePd₃" and perovskite "CeSbPd₃" units. Magnetic measurements indicate an effective moment of $2.45(4)\mu_B$ per cerium with no apparent magnetic order above 3K. Substitution of other large elements of the P-block is underway.

5. Acknowledgements

We would like to acknowledge the support of the Office of Naval Research. We would also like to thank R.L.Gitzendanner and S.S.Trail for assistance with the precession work and Dr. E. Lobkovsky for assistance with the single crystal data collection. We appreciate discussions with Dr. H. Yamane and Dr. N.E. Brese concerning the solving of crystal structures using SHELX-93.

6. References

- 1. F. Steglich, C. Geibel, K. Gloos, G. Olesch, C. Schank, C. Wassilew, A. Loidl, A. Krimmel and G.R. Stewart, J. Low Temp. Physics 95(1/2), 3 (1994).
- 2. M. Loewenhaupt and K.H. Fischer, in: *Handbook on the Chemistry and Physics of Rare Earths*, eds. K.A. Gschneider Jr. and L. Eyring, Chapter 105, Elsevier Science Publishers B.V., 1993.
- 3. J.P. Kappler, M.J. Besnus, P. Lehmann, A. Meyer and J. Sereni, J. Less-Common Met., 111, 261 (1985).
- H. Sthioul, D. Jaccard and J. Sierro, in: Valence Instabilities, eds. P. Wachter and H. Boppart, p443, North-Holland Publishing Company, 1982.
- 5. S.K. Malik, R. Vijayaraghavan, E.B. Boltich, R.S. Craig and W.E. Wallace, Solid State Comm., 43(4), 243 (1982).
- 6. S.K. Dhar, S.K. Malik and R. Vijayaraghavan, Mat. Res. Bull., 16, 1557 (1981).
- 7. G. Nieva, J.G. Sereni and J.P. Kappler, Physica Scripta, 35, 201-203 (1987).
- 8. R.A. Gordon, F.J. DiSalvo and R. Pöttgen, J. Alloys Compounds, accepted.
- 9. S.K. Malik and D.T. Adroja, Phys. Rev. B, 43(7), 6295 (1991).
- 10. W.K. Hofmann and W. Jeitschko, Monatshefte für Chemie, 116, 569 (1985).
- 11. O. Sologub, K. Hiebl, P. Rogl, H. Noël and O. Bodak, J. Alloys Compounds, 210, 153 (1994).
- XSCANS 2.0 Data Collection Package, copyright 1993, 1994, Siemans Industrial Automation Inc.
- 13. P.E. Werner, L. Eriksson and M. Westdahl, J. Appl. Crystallogr., 18, 367 (1985).

- 14. K. Yvon, W. Jeitschko and E. Parthé, J. Appl. Crystallogr., 10, 73 (1977).
- 15. R.A. Gordon, Y. Ijiri, C.M. Spencer and F.J. DiSalvo, J. Alloys Compounds, in press.
- G.M. Sheldrick, SHELX-86, Program for the Solution of Crystal Structures, University of Göttingen, Germany, 1986.
- 17. G.M. Sheldrick, SHELX-93, Program for Crystal Structure Refinement, University of Göttingen, Germany, 1993.
- 18. J. Donohue, The Structures of the Elements, Wiley, New York (1974).
- Structural figures were generated using ATOMS 3.1 (β2) for Windows, copyright 1994,E. Dowty.

Table 1. Crystal data and structure refinement for Ce₈Pd₂₄Sb.

Empirical formula	Ce ₈ Pd ₂₄ Sb
Formula weight	3796.31 g/mol
Temperature	293(2) K
Wavelengths	Μο Κα
Crystal system	cubic
Space group	Pm3m (No. 221)
Unit cell dimensions	a = 8.461(1) Å
	$V = 605.71(2) \text{ Å}^3$
Formula units per cell	Z = 1
Calculated density	10.408 g/cm^3
Crystal size	18 x 38 x 232 μm ³
Absorption correction	from ψ -scan data
Absorption coefficient	33.03 mm ⁻¹
F(000)	1619
θ range for data collection	3° to 60°
Scan mode	ω / θ
Range in hkl	$-1 \leq h, k, l \leq 11$
Total no. reflections	1453
Independent reflections	$222 (R_{int} = 0.0548)$
Refinement method	Full-matrix least-squares on F ²
Data/restraints/parameters	222 / 0 / 16
Goodness-of-fit on F ²	0.959
Final R indices $[I > 2\sigma(I)]$	R1 = 0.0170, wR2 = 0.0412
R indices (all data)	R1 = 0.0216, $wR2 = 0.0412$
Extinction coefficient	0.00648(1)
Largest diff. peak and hole	1.29 and -1.17 e^{-1} Å ³
•	

Table 2. Atomic co-ordinates and isotropic displacement parameters (in pm²) for Ce₈Pd₂₄Sb.

Atom	Wyckoff site	х	у	z	$U_{\rm eq}^{ m a)}$
Ce1	8g	0.25140(3)	0.25140(3)	0.25140(3)	82(2)
Pd1	12h	0.26675(7)	1/2	0	101(2)
Pd2	6e	0.31118(11)	0	0	128(2)
Pd3	6 <i>f</i>	0.25552(8)	1/2	1/2	92(2)
Sb1	1 <i>a</i>	0	0	0	194(4)

 U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table 3. Interatomic distances (Å) in the structure of $Ce_8Pd_{24}Sb$. All distances less than 4.5Å are shown. Standard deviations are typically 0.001Å or less.

Ce1:	3 Pd1 2.975 6 Pd3 2.994 3 Pd2 3.050 1 Sb1 3.684 3 Ce1 4.207 3 Ce1 4.254	Pd1:	4 Pd1 2.925 4 Pd3 2.927 4 Ce1 2.975 1 Pd1 4.137 4 Pd3 4.232	Pd3:	2 Pd2 2.765 2 Pd3 2.791 2 Pd1 2.927 4 Ce1 2.994 2 Pd3 3.192 1 Pd3 3.947
Sb1:	6 Pd2 2.633 8 Ce1 3.684	Pd2:	1 Sb1 2.633 4 Pd3 2.765 4 Ce1 3.050 1 Pd2 3.195 4 Pd2 3.724 4 Pd3 4.247		2 Pd1 4.232

Table 4. Comparison of observed and calculated powder diffraction intensities for Ce₈Pd₂₄Sb.

		0 24		
h k l	d _{obs.} (Å)	d _{calc.} (Å)	I _{obs.} (%)	I _{calc} (%)
2 2 0	2.9844	2.9858	7	4.7
300	2.8139	2.8150	3	2.6
2 2 1		2.8150		0.4
3 1 0	2.6690	2.6705	1	1.7
222	2.4369	2.4379	100	100
3 2 0	2.3404	2.3422	7	6.3
3 2 1	2.2568	2.2570	4	3.3
400	2.1106	2.1112	48	55
4 1 0		2.0482		0.9
3 2 2	2.0474	2.0482	8	6.3
3 3 0	1.9899	1.9905	2	1.7
4 1 1		1.9905		0.6
4 2 0	1.8885	1.8884	1	0.9
3 3 2	1.8007	1.8005	2	3.0
4.3 2	1.5677	1.5682	1	1.2
5 2 0		1.5682		0.4
4 4 0	1.4927	1.4929	40	31
5 2 2		1.4701		0.4
4 4 1	1.4696	1.4701	1	1.0
600		1.4075		0.2
4 4 2	1.4079	1.4075	1	1.3
620	1.3347	1.3353	2	3.0
5 4 0	1.3185	1.3189	2	1.3
6 2 1		1.3189		1.2
4 4 3		1.3189		0.3
6 2 2	1.2727	1.2731	27	30
630	1.2584	1.2589	3	2.2
5 4 2		1.2589		1.2
4 4 4	1.2186	1.2189	9	9.3
632	1.2063	1.2064	3	5.0
700		1.2064		0.3

Figure Captions

- Fig. 1. Crystal structure of Ce₈Pd₂₄Sb [19]. Palladium to palladium connections are shown to emphasize distortions in the structure. In order of decreasing size, circles represent: Ce (white), Sb (black) and Pd (gray).
- Fig. 2. Distorted (a) and undistorted (b) sub-cells of Ce₈Pd₂₄Sb. Atom designations are as indicated in Fig. 1.
- Fig. 3. Inverse magnetic susceptibility as a function of temperature from 3K to 320K for $Ce_8Pd_{24}Sb$. Data from 3K to 50K is shown inset.

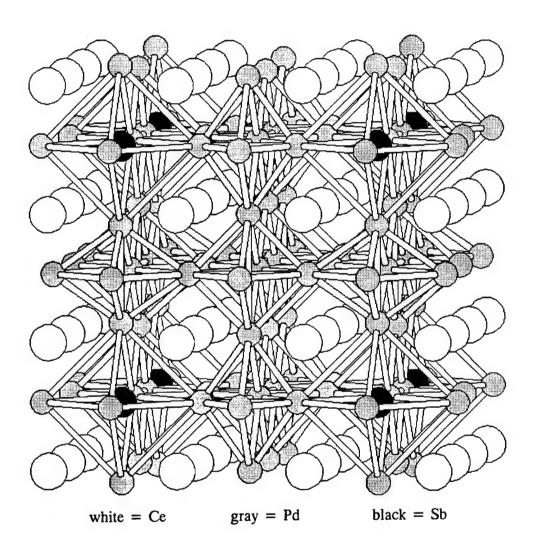
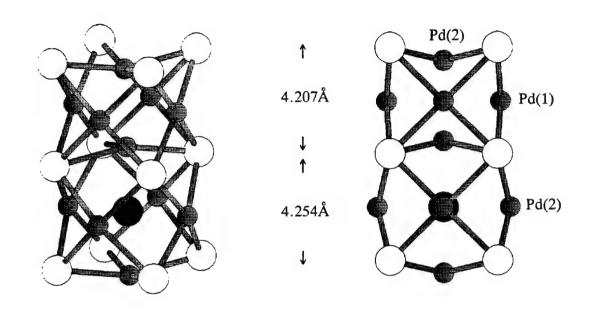
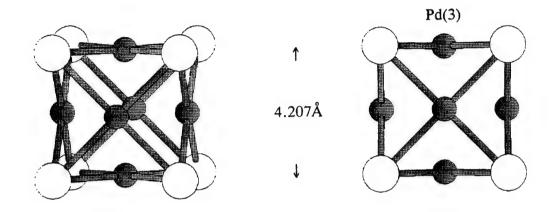


figure 1





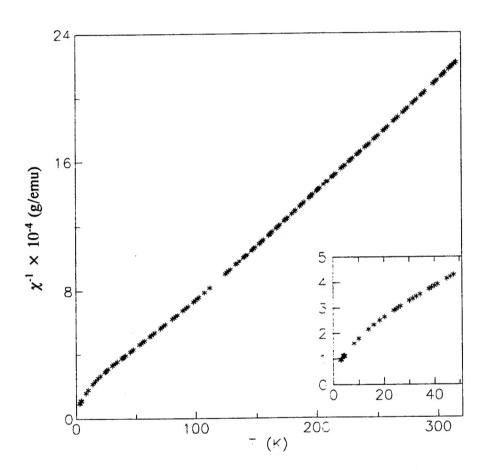


figure 3

10s

146558211897996110875841114242410289498989189142156128512114

2178211054 9646661983097 278864115148217316338881861211811316

L 10Fo 10Fc 10s l 10Fo 10Fc h k L 10Fo 10Fc 10s h k 10Fo 10Fc 10s k h 340123001012012301234012340120010120123012 66777701122233334444455555666601122233333444 935 247 1 1445 604 527 95 527 2 1445 604 433 305 527 2 152 647 362 2 152 647 3 152 647 647 4 165 646 2 10590220242985545632224363777142716044007599286669 625 9100 9107 1559 1559 2318 651 1664 1664 1748 1813 6677 11003 1846 1003 18 234501234560123450123001012012301234012345012 5555666666677777788888011122233333444445555555666 6157804456075470218051628117007722303094551707261 123401234500101201230123401234501234560010120 44445555555011222333334444445555556666666601122223 123012340123450123456012345600101201230123401 33344444455555566666666777777770112223333344444455 3911231177702158684093353264589761690853889726517

Table 2. Anisotropic displacement parameters (in pm²) for Ce₈Pd₂₄Sb.

АТОМ	U11	U22	U33	U23	U13	U12
Ce(1) Pd(1) Pd(2) Pd(3) Sb(1)	82(2)	82(2)	82(2)	2(1)	2(1)	2(1)
	115(2)	82(3)	106(2)	0	0	0
	211(5)	87(2)	87(2)	0	0	0
	104(3)	87(2)	87(2)	0	0	0
	194(4)	194(4)	194(4)	0	0	0

800 N. Quincy St. Dr. John C. Pazik Arlington, VA 22217-5660 Physical S&T Division - ONR 331 Office of Naval Research Chemistry Division, Code 385 Defense Technical Information NAWCWD - China Lake Ctr (2) China Lake, CA 93555-6001 Building 5, Cameron Station Alexandria, VA 22314 Dr. Peter Seligman (1) NCCOSC - NRAD Dr. James S. Murday San Diego, CA 92152-5000 (1)Chemistry Division, NRL 6100 Dr. Bernard E. Douda (1) Naval Research Laboratory Crane Division Washington, DC 20375-5660 **NAWC** Crane, Indiana 47522-5000 Dr. John Fischer (1) * Number of copies required

******** page 6**

Technical Report Distribution List